University of Louisville College of Arts and Sciences

# Department of Physics and Astronomy PhD Qualifying Examination (Part I)

# Fall 2014

Paper C – Thermal Physics

Time allowed – 90 minutes

### **Instructions and Information:**

- Answer both questions
- This is a closed book examination
- Start each question on a new sheet of paper use only one side of each sheet
- Write your identification number on the upper right hand corner of each answer sheet
- You may use a non programmable calculator
- Partial credit will be awarded.
- Correct answers without adequate explanations will not receive full credit.
- Make sure your work is legible and clear
- The points assigned to each part of each question is clearly indicated

### **Thermal Physics Basic**

A steel ball bearing, radius 2 cm, volume 2.5 x  $10^{-5}$  m<sup>3</sup>, initially at 4  $^{0}$ C is dropped into a tank containing 2.5 kg of liquid Nitrogen initially at 73 K. The equilibrium temperature is exactly the boiling point of liquid Nitrogen (77 K), but none of the liquid Nitrogen boils away. (Latent heat of vaporization of Nitrogen = 200,000 J/kg; specific heat of liquid nitrogen = 2000 J/kg.K; specific heat of steel = 500 J/kg.K; specific heat of Nitrogen gas = 750 J/kg.K; 0  $^{0}$ C = 273 K).

#### In all situations you should assume that no heat is gained or lost from the surroundings.

(a)	How many joules of heat are released when the temperature of 1 kg of steel drops by 1	°C ? (2)
(b)	Determine the mass of the ball bearing.	(8)
(c)	If the volume coefficient of thermal expansion of steel is $36 \ge 10^{-6} \text{ K}^{-1}$ , calculate the chavelenge of the ball bearing.	inge in (7)
(d)	Determine the change in the radius of the ball bearing.	(7)
(e)	A second steel ball bearing, mass 5 kg, is dropped into a different tank containing the sa	ime ma

(e) A second steel ball bearing, mass 5 kg, is dropped into a different tank containing the same mass of liquid Nitrogen at the same initial temperature (73K). Determine the equilibrium temperature in this case. (11)

#### **Thermal Physics Intermediate**

The statistics of paramagnetism can be studied using a model system of N identical, localized, noninteracting magnetic dipoles, each with a dipole moment of m, in a uniform applied magnetic field H. We consider here the case of classical dipoles that can be oriented in any direction relative to the applied field. In such a case, the total energy of the system can be written as

$$E = \sum_{i=1}^{N} E_i = -\sum_{i=1}^{N} \vec{\mu}_i \cdot \vec{H} = -\mu H \sum_{i=1}^{N} \cos \theta_i$$

Assume the applied magnetic field is in the positive z direction, the angle  $\theta_i$  would be the angle between the magnetic moment of the i-th dipole and the z axis.

(a)	Calculate the partition function of the system at a given temperature T.	(25)
(b)	Calculate the mean magnetic moment of the system.	(20)

(c) Show that in the high temperature limit,  $\frac{k_B T}{\mu H} \gg 1$ , the mean magnetic moment of the whole system is approximately  $CN \frac{H}{T}$  (*this is the so - called Curie law*), and find the constant C. (20)